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ARPES study of Fe_xTiS₂ ($x = 0 \le x \le 0.33$)

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1T-TiS₂ is non-magnetic layered material with 1T-CdI₂-type crystal structure. The hexagonal layer of Ti ions is sandwiched between the two hexagonal layers of S ions and the Ti ion is octahedrally coordinated with six S ions. The S-Ti-S triple layers are covalently bonded and these TiS2 triple layers are weakly coupled with van der Waals (vdW) force. In the vdW gap located between the TiS₂ layers, the other 3d transitionmetal M can be intercalated as $M_x TiS_2$. Among them, $Fe_x TiS_2$ exhibits a wide variety of magnetic properties [1]. With increasing the Fe concentration from x = 0, the cluster spin grass (CG) state with the Ising spins is found for x < 0.20. After exhibiting the antiferromagnetic (AFM) state from x = 0.20 to 0.28, the CG state is again realized for 0.28 < x < 0.38 and the AFM state for 0.38 < x < 0.50. Above x = 0.50, the ferrimagnetic behavior is observed. On the other hand, the Fe ion intercalated in the vdW gap between the TiS2 layers occupy the octahedral site surrounded by six S ions. X-ray studies reveal the Fe random distribution for x < 0.20, while the formation of $2\sqrt{3}a \times 2a \times 2c$ superlattices due to the Fe ordering for x = 0.25 and $\sqrt{3}a \times \sqrt{3}a$ \times 2c superlattice for x = 0.33. The rich magnetic states of Fe_xTiS₂ are expected to link to the change in electronic band structure due to the Fe intercalation. In this study, we carried out angle-resolved photoemission spectroscopy (ARPES) to reveal the electronic band structure of Fe_xTiS₂ ($0 \le x \le 0.33$) at beamlines BL-1, BL-7 and BL-9A of Hiroshima Research Institute for Synchrotron Radiation Science (HiSOR).

Figures 1(a), (b) and (c) shows the ARPES intensity plots of TiS₂, Fe_{0.25}TiS₂ and Fe_{0.33}TiS₂, respectively, measured at hv=66 eV along the $\bar{\Gamma}$ - \bar{M} directions of the surface Brillouin zone, roughly corresponding to Γ-M direction of the bulk Brillouin zone, estimated from the hv-dependent measurements. The ARPES spectra were taken at 20 K and with p-polarized geometry. For TiS₂ in Fig. 1(a), we find an electron pocket derived from the Ti 3d states around the M point, which is resonantly enhanced in intensity around the Ti 3p-3d absorption region. Almost non-dispersive band is observed just below the electron pocket, which is assigned to the localized 3d states of the Ti ions self-intercalated in the vdW gap between TiS₂ layers. The electron pocket is hardly observed in the ARPES spectra with s-polarized geometry (not shown here), suggesting that it comes from the Ti $3d_{z2}$ component. The parabolic band around the Γ point is due to the S 3p states.

The electron pocket around the M point is shifted to higher binding energy on going from TiS₂ (Fig. 1(a)) to Fe_{0.25}TiS₂ (Fig. 1(b)) and further to Fe_{0.33}TiS₂ (Fig. 1(c)) and as the results it becomes large. In addition, the parabolic S 3p band also becomes deeper with increasing the Fe concentration. These results indicate that electrons are transferred from the intercalated Fe ions to host TiS₂ layers. The dispersive Fe 3d-derived band is observed around $E_B = 0.4$ eV for Fe_{0.25}TiS₂ and Fe_{0.33}TiS₂ as shown by dashed lines in Figs. 1(b) and 1(c), reflecting the Fe ions align periodically in the vdW gap parallel to the TiS₂ layers (in plane). The periodicity of the Fe 3d band for Fe_{0.25}TiS₂ becomes half of that of $\overline{\Gamma}$ - \overline{M} - $\overline{\Gamma}$ for TiS₂, reflecting the $2\sqrt{3}a \times 2a$ periodicity in plane for x = 0.25. The periodicity of the Fe 3d band becomes long for Fe_{0.33}TiS₂, which is qualitatively understood from the shorter $\sqrt{3}a \times \sqrt{3}a$ periodicity in plane for x = 0.33. In addition, the Fe 3d band is more dispersive for Fe_{0.33}TiS₂. On the other hand, the Fe 3d band is almost flat for Fe_{0.10}TiS₂ and

 $Fe_{0.15}TiS_2$, indicating that the Fe 3d states almost localized due to the random distribution of the Fe ions.

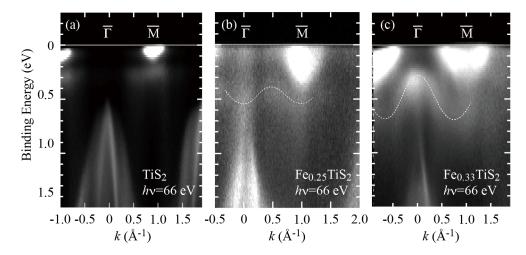


FIGURE 1. ARPES intensity plots of (a) TiS₂, (b) Fe_{0.25}TiS₂ and (c) Fe_{0.33}TiS₂ measured at $h\nu = 66$ eV along $\bar{\Gamma}$ - \bar{M} direction. Dashed lines in (b) and (c) are guide for the eye for the Fe 3d band.

We measured the ARPES spectra of TiS₂, Fe_{0.25}TiS₂ and Fe_{0.33}TiS₂ along the Γ -A direction with changing hv from 30 to 124 eV at 20 K with p-polarized geometry. We find that the Fe 3d bands for Fe_{0.25}TiS₂ and Fe_{0.33}TiS₂ exhibit the dispersion along the direction normal to the TiS₂ layers and has half period of the S 3p band, reflecting the 2c periodicity for x = 0.25 and 0.33.

REFERENCES

1. N. V. Selezneva, E. M. Sherokalova, A. Podlesnyak, M. Frontzek and N. V. Baranov, Phys. Rev. Mater. 7, 014401 (2023).